## ATOMIC ENERGY CENTRAL SCHOOL-3, TARAPUR

## Work sheet MODULE-4

## UNIT-p-BLOCK ELEMENTS CHAPTER: 18-GROUP ELEMENTS

Answer the following questions:

- 1. Group -18 elements are called as noble gases. Why?
- 2. What prompted Bartlett for the discovery noble gas compounds?
- 3. Noble gases have very low boiling points. Why?
- 4. Does the hydrolysis of XeF<sub>4</sub> lead to a redox reaction?
- 5. Xenon does not fluorides such as XeF<sub>3</sub> and XeF<sub>5</sub>. Explain.
- 6. Why do noble gases form compounds with fluorine and oxygen?
- 7. Why is helium used in diving apparatus?
- 8. Why has it been difficult to study the chemistry of radon?
- 9. Which of the following does not exist and why?
- (i) XeOF<sub>4</sub> (ii) NeF<sub>2</sub> (iii) XeF<sub>2</sub> (iv) XeF<sub>2</sub>
- 10. Balance the following reaction:
- (i)  $XeF_4 + H_2O ----> Xe + XeO_3 + HF + O_2$
- (ii)  $XeF_6 + H_2O ----> XeO_2F_2 + HF$

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